

# Project Overview

## *Laser Spectroscopy Group*

<http://laser-spectroscopy.ucc.ie>

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# **Project 1 & 2**

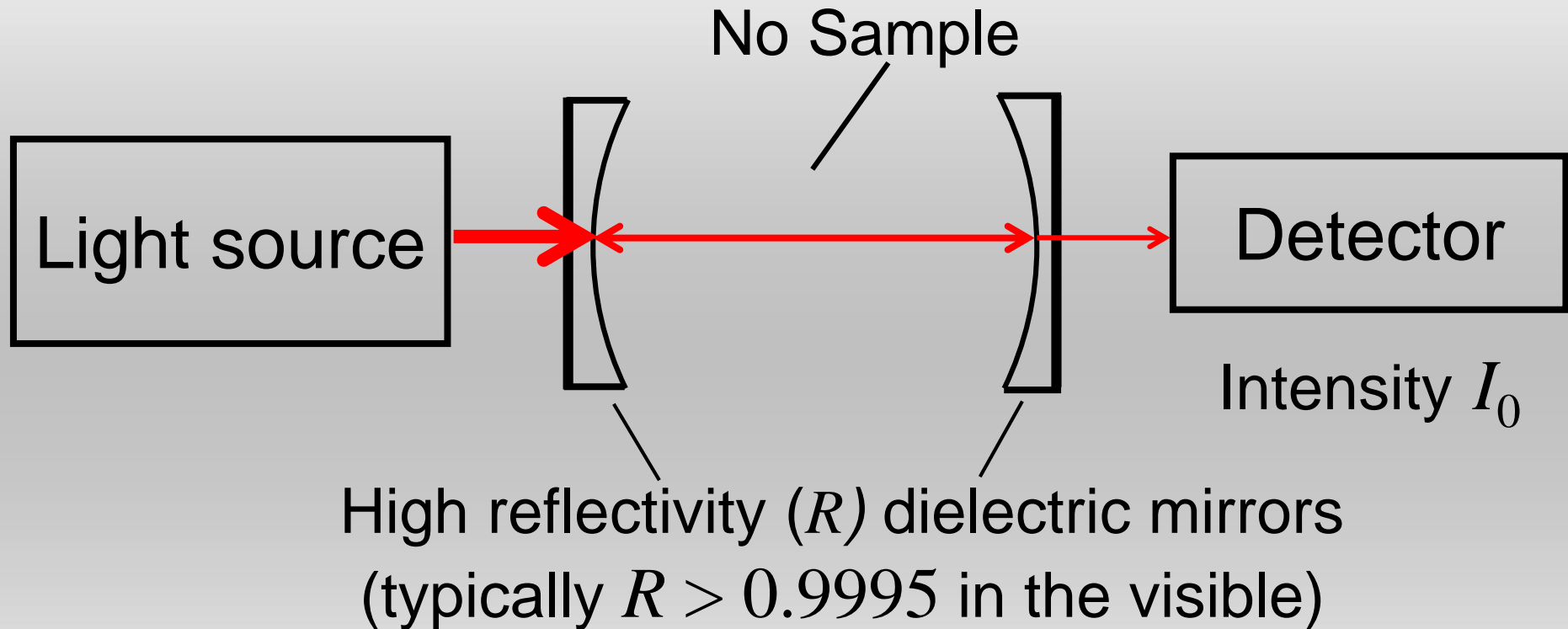
## **Incoherent Broadband Cavity Enhanced Absorption Spectroscopy**

- (1) A compact instrument for NO<sub>2</sub> detection**
- (2) Atmospheric simulation experiments**

# Desirable features of a spectroscopic absorption experiment?

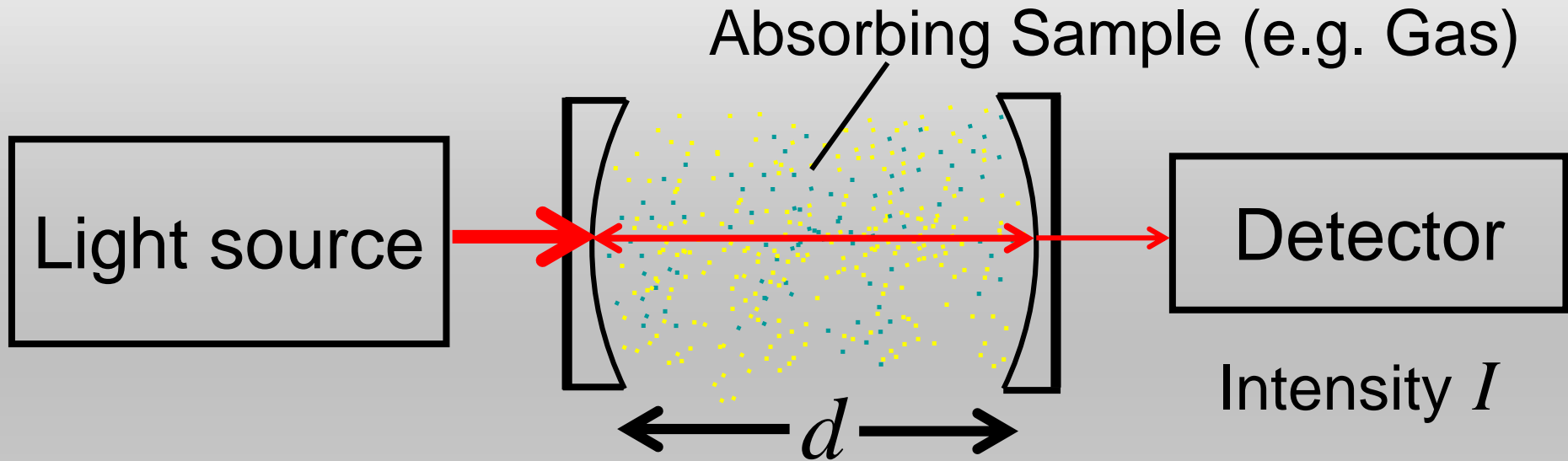
- **Sensitivity**      long (eff.) absorption path length
- **Selectivity**      unambiguous species identification
- **Speed**      high time resolution
- **Quantitative and Direct Methodology**
- **Simplicity / Robustness / Reliability**
- **Versatility**

# Cavity-enhanced absorption spectroscopy



Depending on the mirror reflectivity **several kilometer path lengths** can be achieved !

# Cavity-enhanced absorption spectroscopy



Absorption coefficient  $\alpha$ :

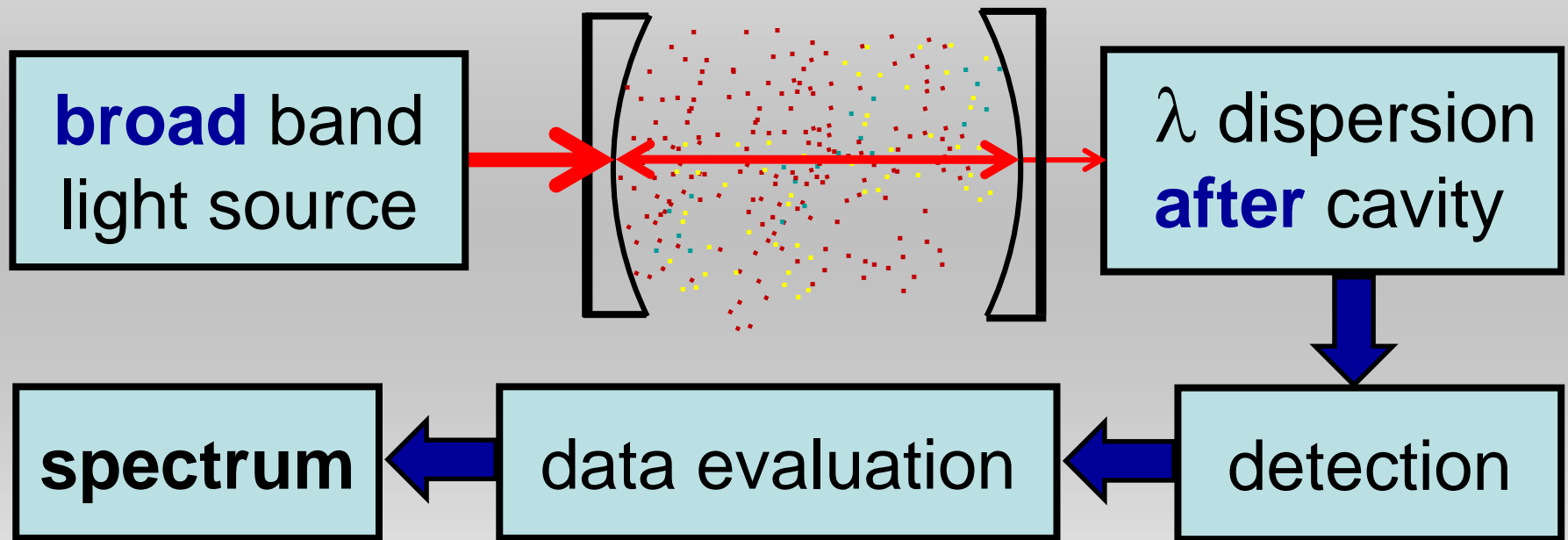
$$\alpha(\lambda) \approx \frac{1}{d} \left( \frac{I_0(\lambda)}{I(\lambda)} - 1 \right) [1 - R(\lambda)]$$

# Broadband Cavity-Enhanced Methods

## Measurement principle:

(A) Spectrally **broad** light coupled into cavity

(B) Dispersion of wavelength **after** the cavity

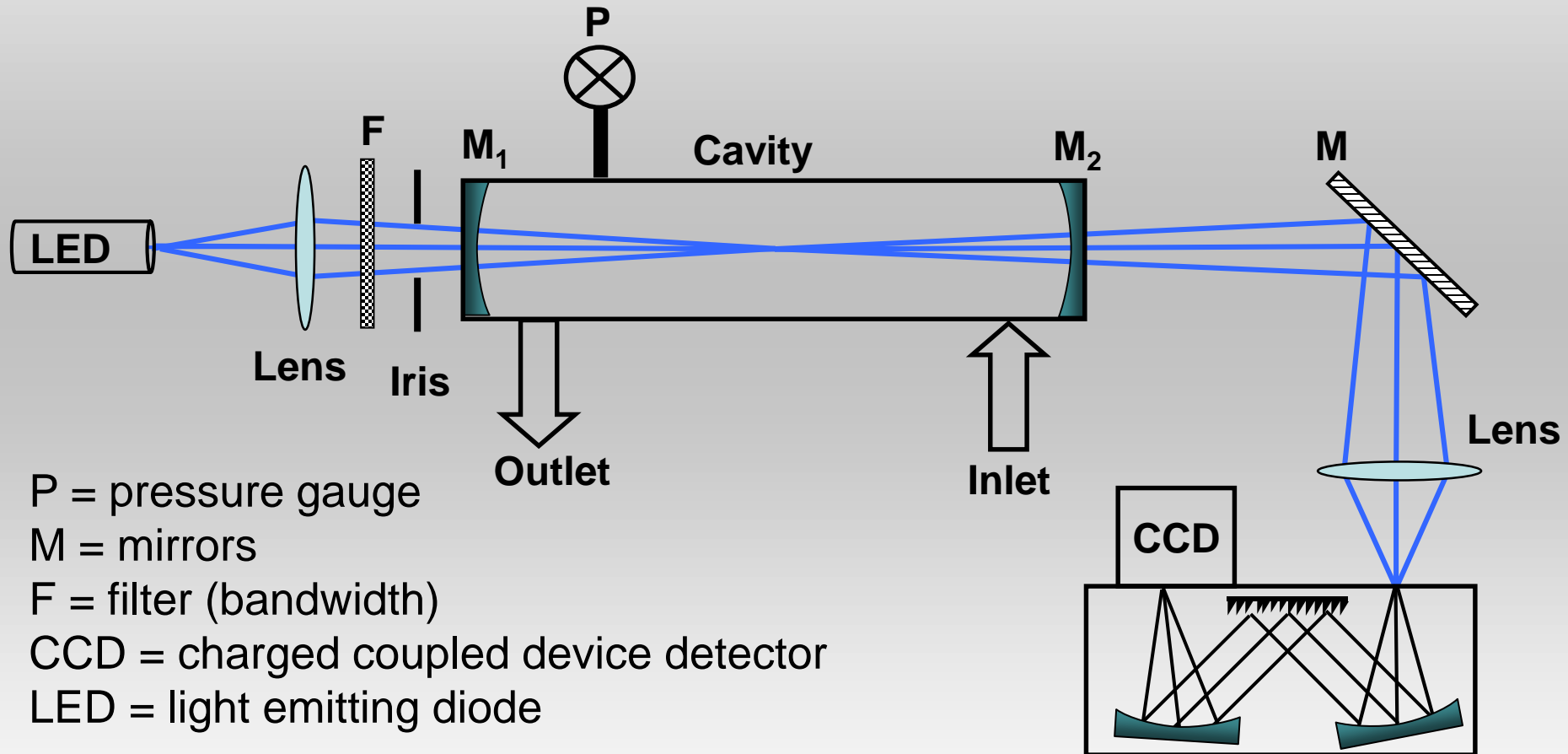


## Multiplexing advantage:

(A) No scanning of wavelength required (in principle)

(B) High time resolution for wide spectral ranges

# (1) A compact instrument for $\text{NO}_2$ detection

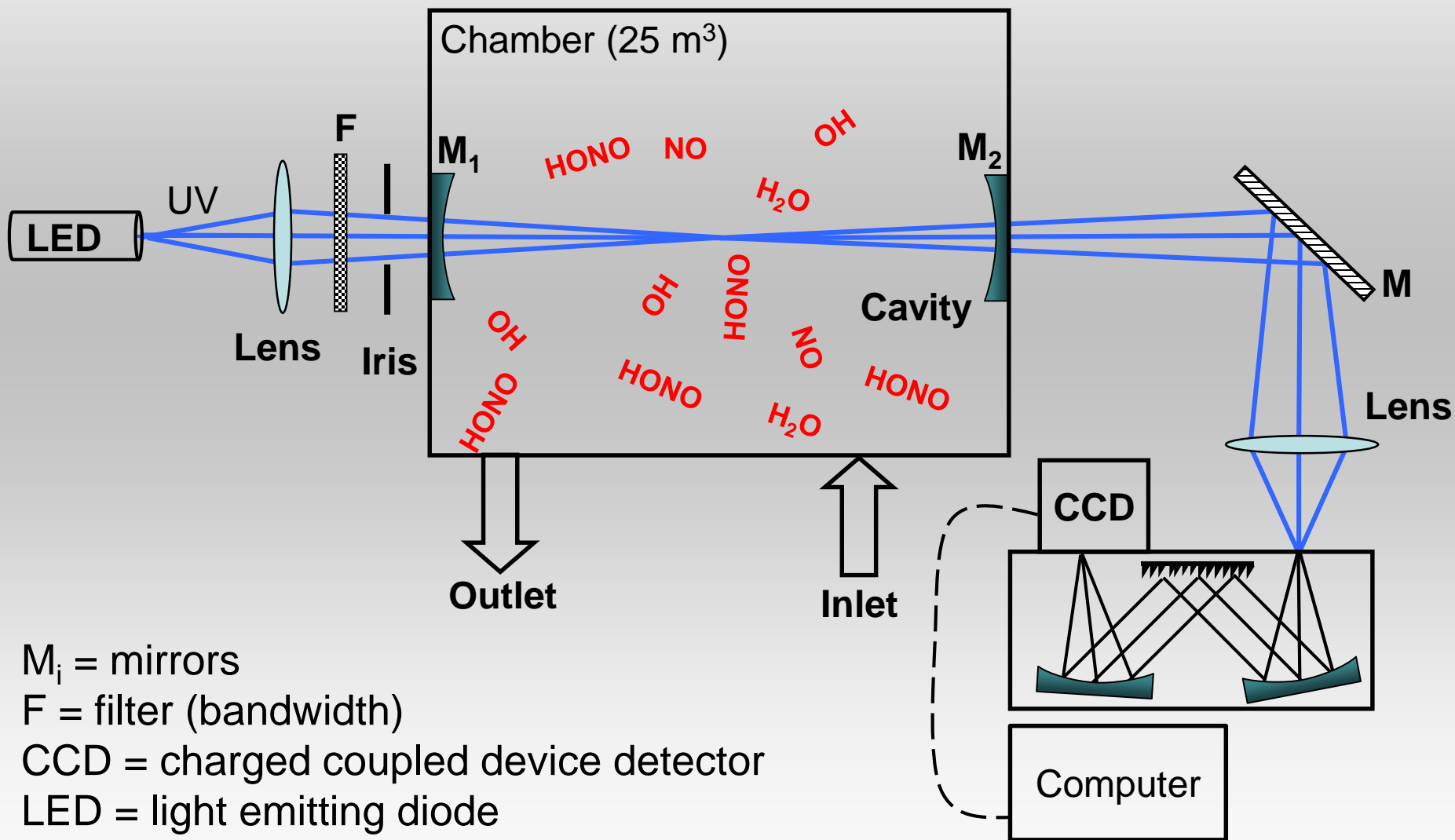


# Objectives / Activities (1)

- Characterization of blue LED.  
(spectrum, brightness, power, stability, divergence...)
- Setting up of IBBCEAS apparatus.  
(detection system – CCD, optical cavity, filter constraints)
- Proof-of-principle experiments. (characterization of experimental parameters: signal-to-noise ratio, detection limit, Allan variance...)
- Mirror reflectivity calibration.
- Setting up gas system.
  - (a) close gas cell test measurements with NO<sub>2</sub>
  - (b) open path test measurements in ambient air



## (2) Atmospheric simulation experiments



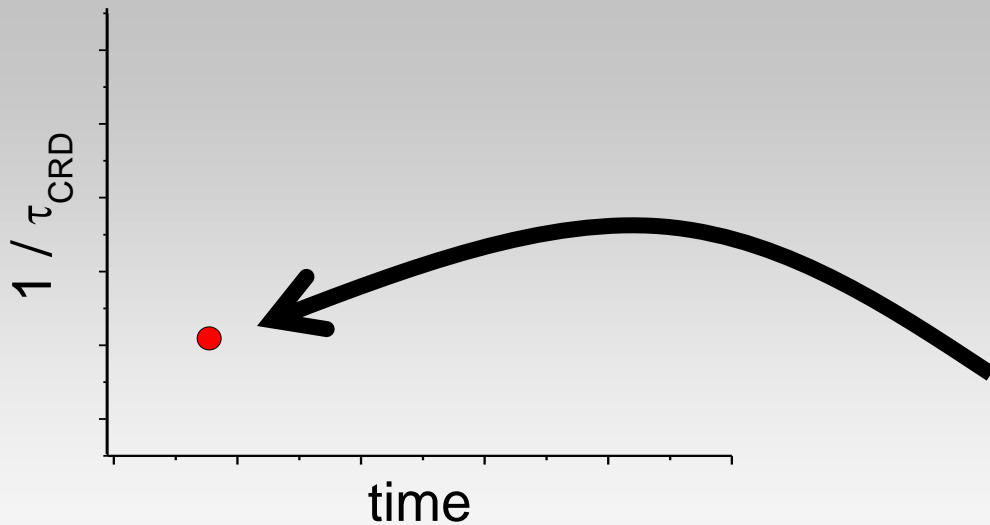
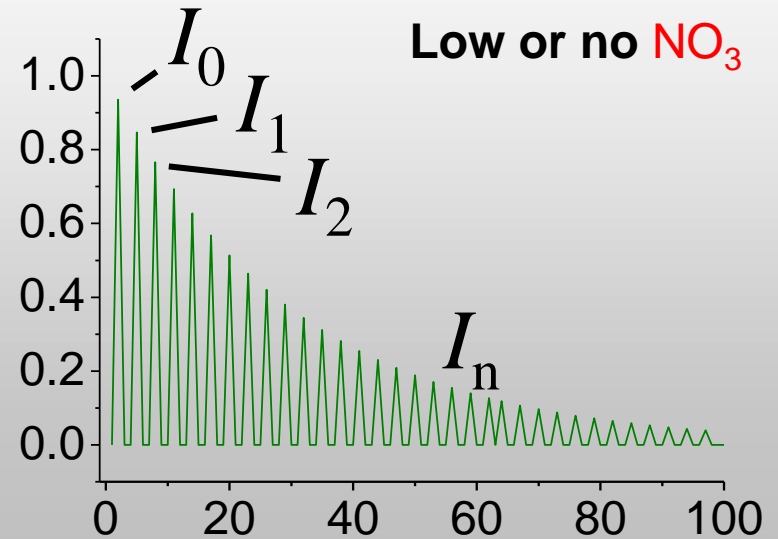
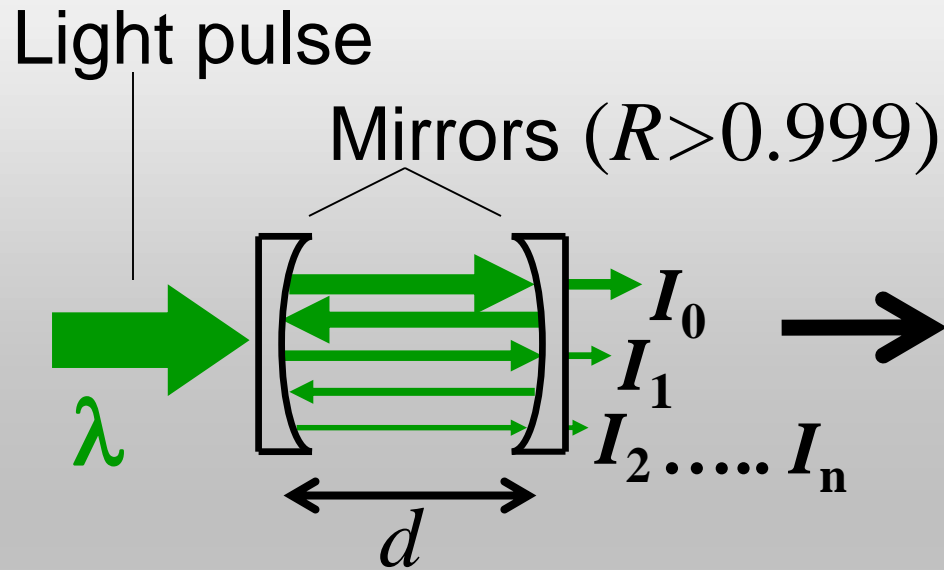
# Objectives / Activities (2)

- Characterization of UV LED.  
(spectrum, brightness, power, stability, divergence...)
- Setting up of IBBCEAS apparatus, implementation on atmospheric simulation chamber  
(detection system – CCD, optical cavity, filter constraints)
- Proof-of-principle experiments. (characterization of experimental parameters: signal-to-noise ratio, detection limit, Allan variance...)
- Mirror reflectivity calibration.
- Measurement and discussion of:
  - (a) open path HONO spectra
  - (b) open path NO<sub>2</sub> Spectra and comparison with commercial chemiluminescence detector

# **Project 3**

**Trace gas absorption measured  
by cavity ring-down  
spectroscopy**

# Principle of the cavity ring-down method

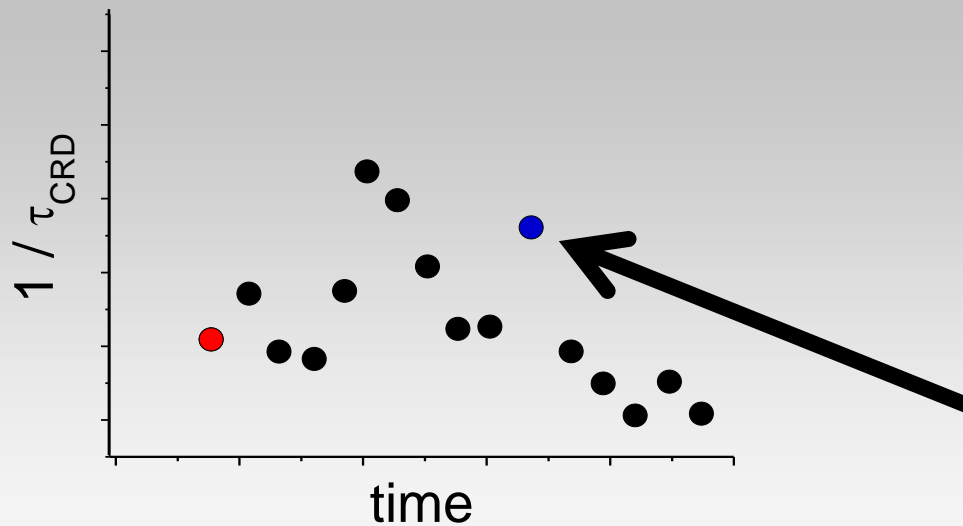
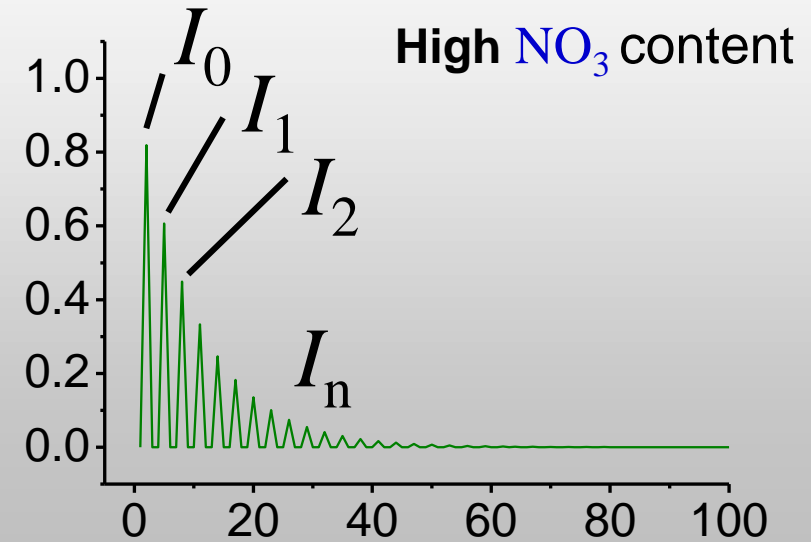
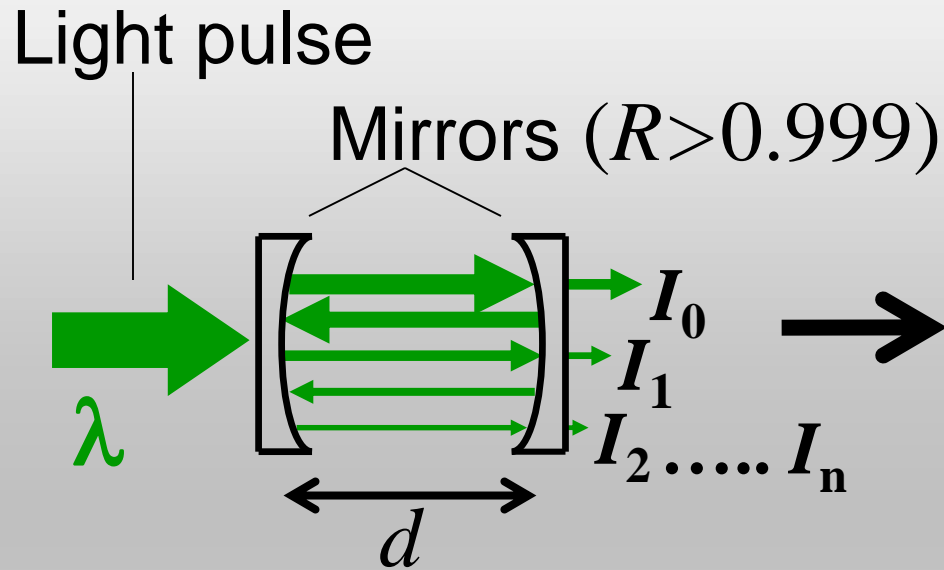


fit

$$I(t) = I_0 \exp\left(-\frac{t}{\tau_{\text{crd}}}\right)$$

$$\tau_{\text{crd}}^{-1} = \frac{(1-R)c}{d} + \underline{c\sigma n_{\text{NO}_3}}$$

# Principle of the cavity ring-down method

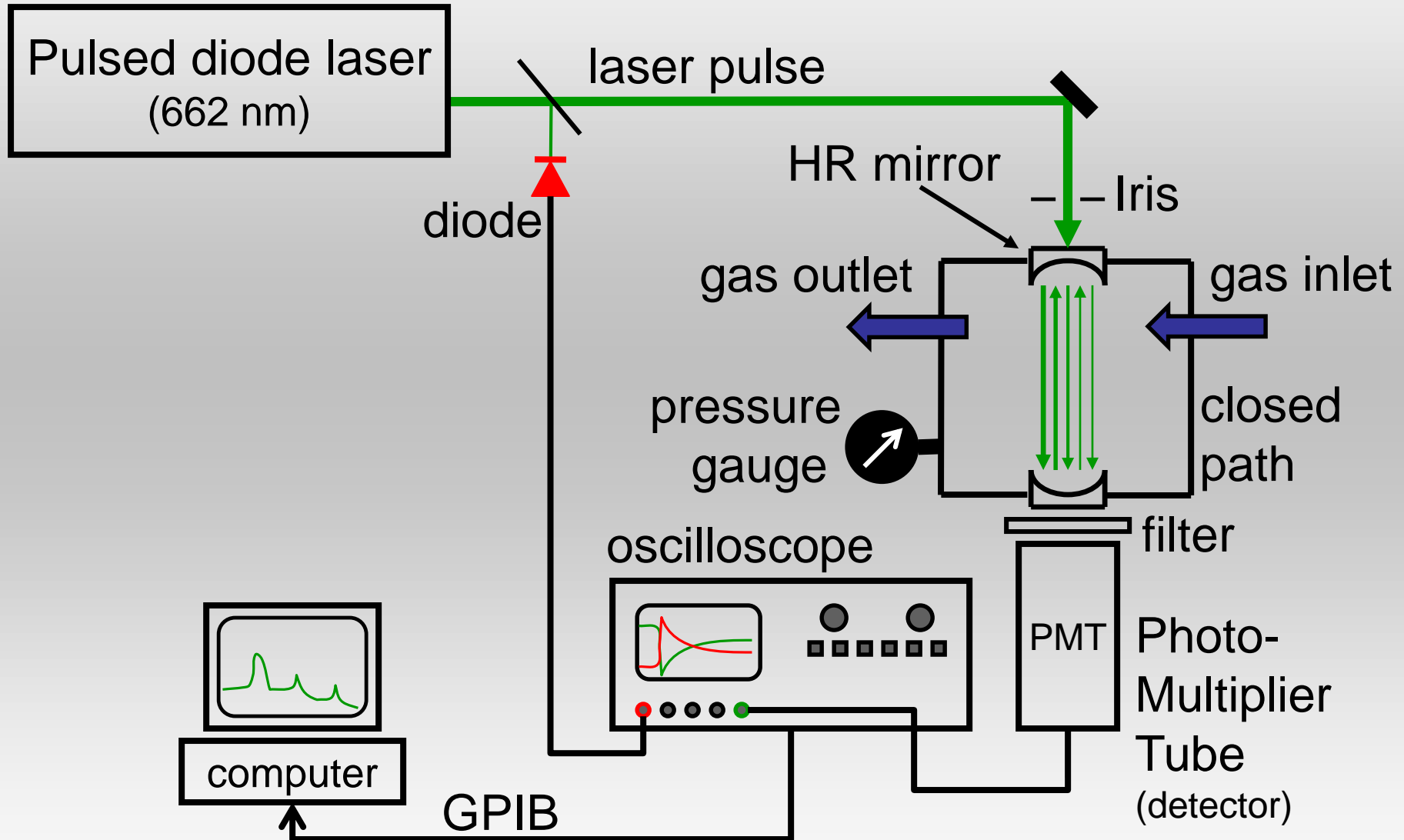


fit

$$I(t) = I_0 \exp\left(-\frac{t}{\tau_{\text{crd}}}\right)$$

$$\tau_{\text{crd}}^{-1} = \frac{(1-R)c}{d} + \underline{c\sigma n_{\text{NO}_3}}$$

# Cavity Ring-Down Experiment



# Objectives / Activities (3)

- Setup a vacuum tight, pressure controlled, *closed-path* cavity.
- Setup software for measurement and data analysis.
- Measure ring-down times in the empty cavity to characterize the optical losses (e.g. as a function of pressure).
- Detect H<sub>2</sub>O absorption in the cavity as a function of time (optimize integration time).
- Synthesize NO<sub>3</sub> (=NO<sub>2</sub> + O<sub>3</sub>) in the cavity and detect the nitrate radical concentrations as a function of time (optimize integration time).