

Project Overview

Laser Spectroscopy Group

<http://laser-spectroscopy.ucc.ie>

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Project 1 & 2

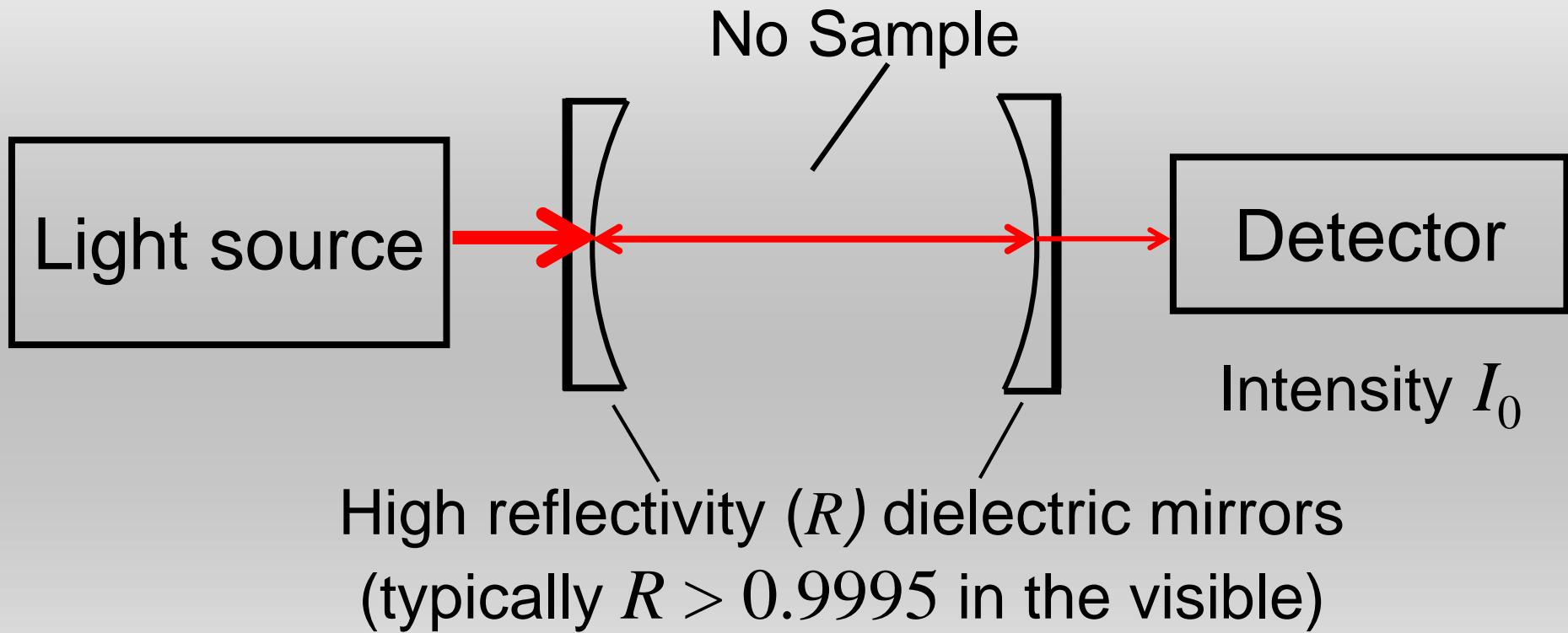
Incoherent Broadband Cavity Enhanced Absorption Spectroscopy

- (1) A compact instrument for NO₂ detection
- (2) Atmospheric simulation experiments

Desirable features of a spectroscopic absorption experiment?

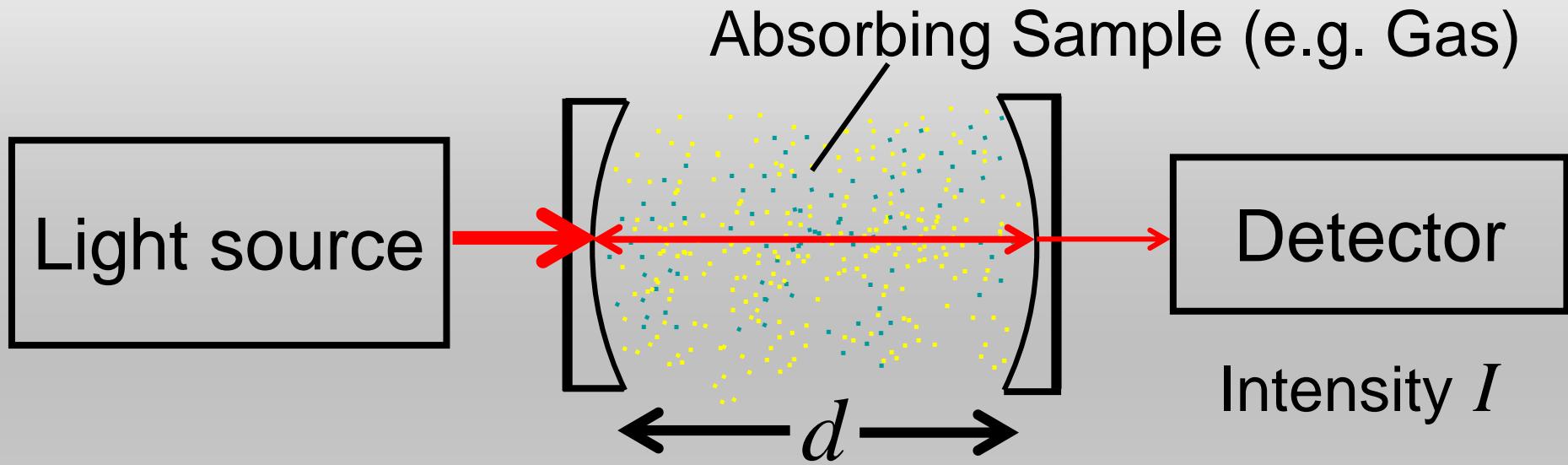
- **Sensitivity** long (eff.) absorption path length
- **Selectivity** unambiguous species identification
- **Speed** high time resolution
- **Quantitative and Direct Methodology**
- **Simplicity / Robustness / Reliability**
- **Versatility**

Cavity-enhanced absorption spectroscopy



Depending on the mirror reflectivity several kilometer path lengths can be achieved !

Cavity-enhanced absorption spectroscopy



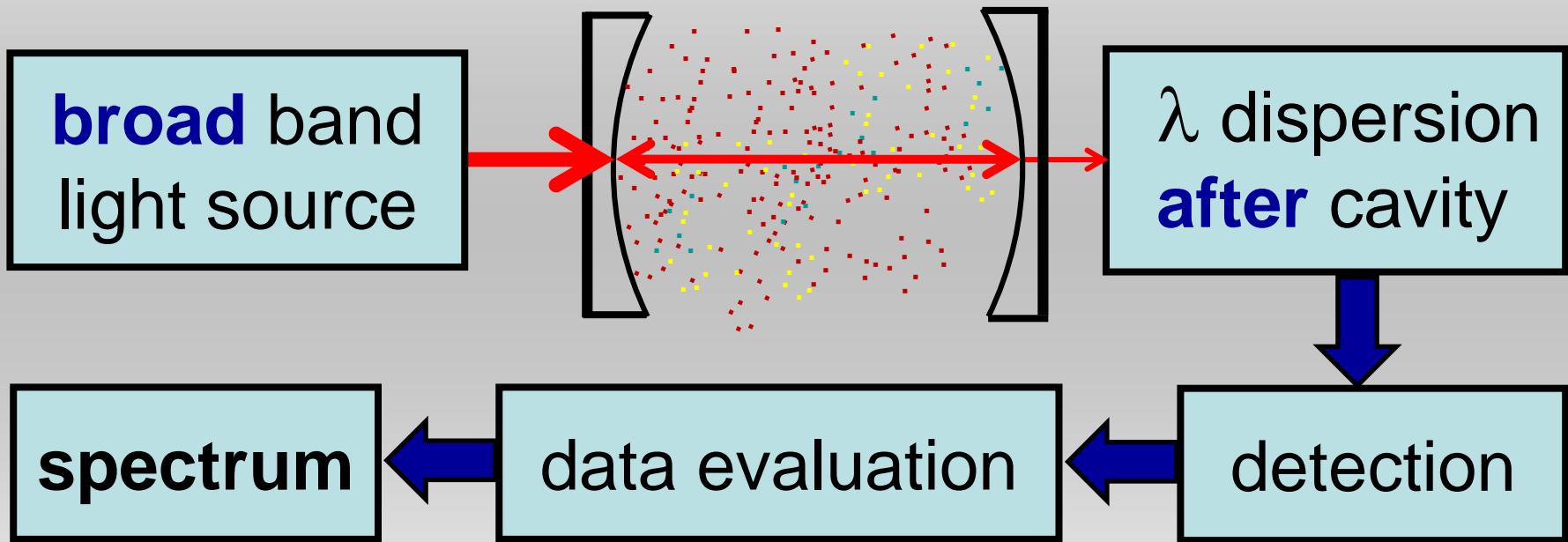
Absorption coefficient α :

$$\alpha(\lambda) \approx \frac{1}{d} \left(\frac{I_0(\lambda)}{I(\lambda)} - 1 \right) [1 - R(\lambda)]$$

Broadband Cavity-Enhanced Methods

Measurement principle:

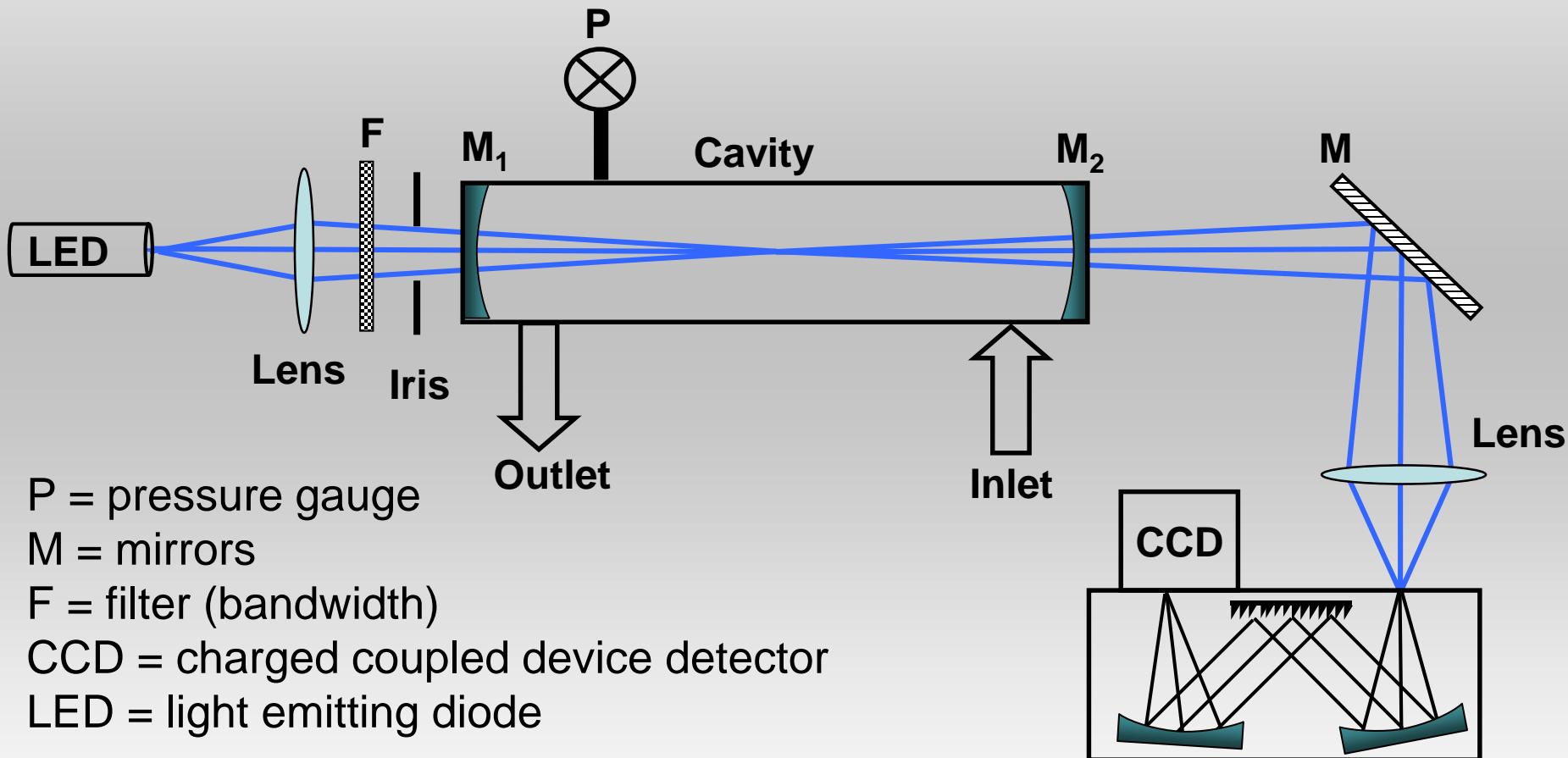
- (A) Spectrally **broad** light coupled into cavity
- (B) Dispersion of wavelength **after** the cavity



Multiplexing advantage:

- (A) No scanning of wavelength required (in principle)
- (B) High time resolution for wide spectral ranges

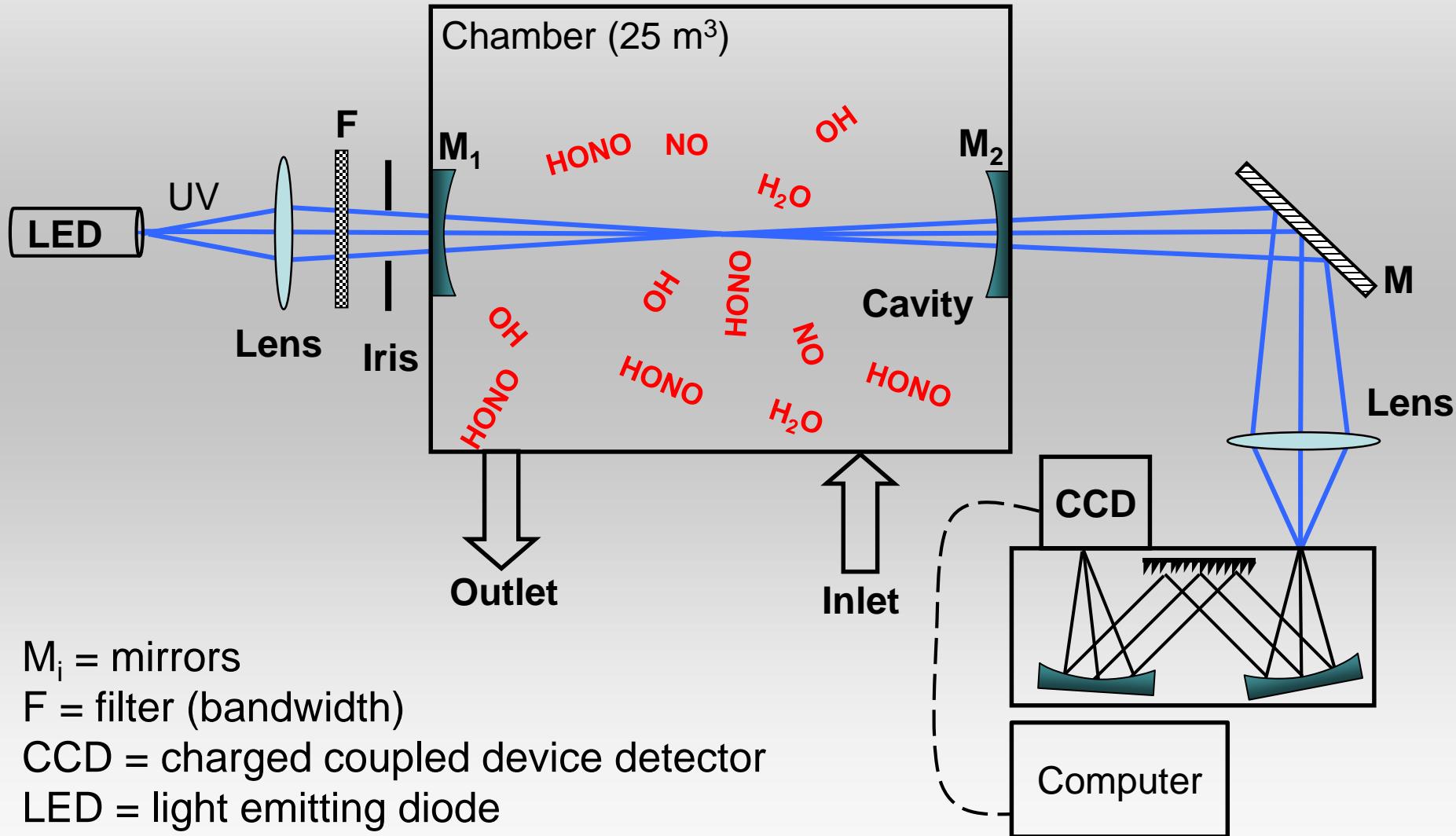
(1) A compact instrument for NO₂ detection



Objectives / Activities (1)

- Characterization of blue LED.
(spectrum, brightness, power, stability, divergence...)
- Setting up of IBBCEAS apparatus.
(detection system – CCD, optical cavity, filter constraints)
- Proof-of-principle experiments. (characterization of experimental parameters: signal-to-noise ratio, detection limit, Allan variance...)
- Mirror reflectivity calibration.
- Setting up gas system.
 - (a) close gas cell test measurements with NO_2
 - (b) open path test measurements in ambient air

(2) Atmospheric simulation experiments



Objectives / Activities (2)

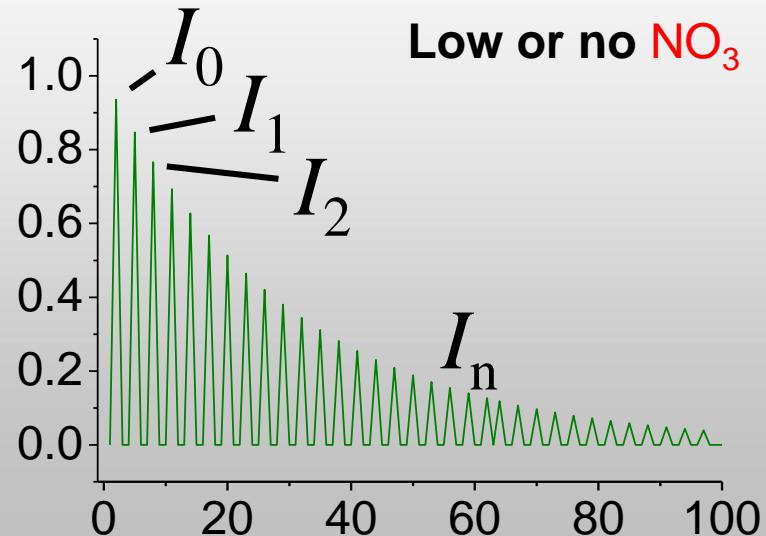
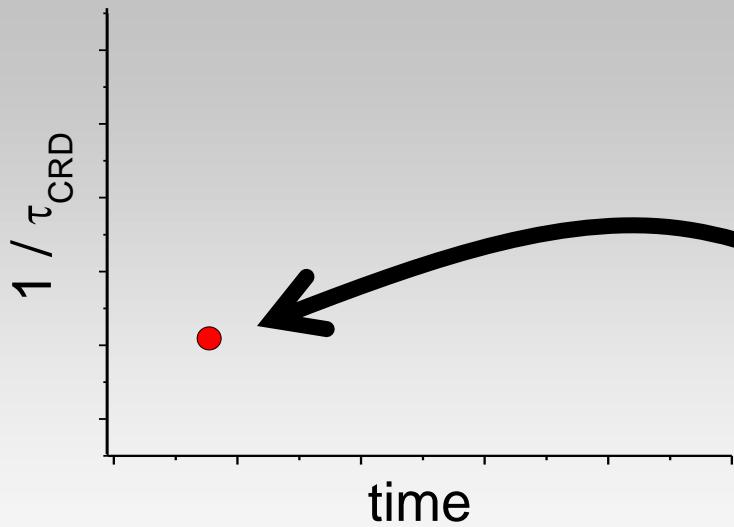
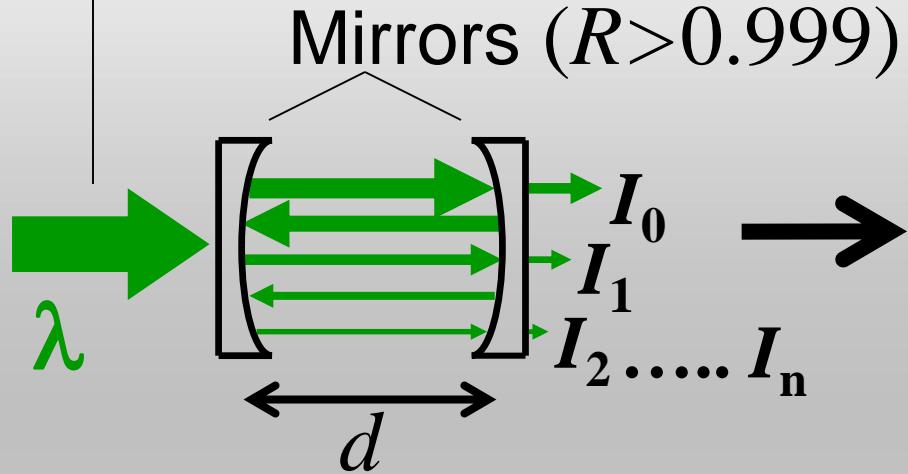
- Characterization of UV LED.
(spectrum, brightness, power, stability, divergence...)
- Setting up of IBBCEAS apparatus, implementation on atmospheric simulation chamber
(detection system – CCD, optical cavity, filter constraints)
- Proof-of-principle experiments. (characterization of experimental parameters: signal-to-noise ratio, detection limit, Allan variance...)
- Mirror reflectivity calibration.
- Measurement and discussion of:
 - (a) open path HONO spectra
 - (b) open path NO_2 Spectra and comparison with commercial chemiluminescence detector

Project 3

Trace gas absorption measured
by cavity ring-down
spectroscopy

Principle of the cavity ring-down method

Light pulse



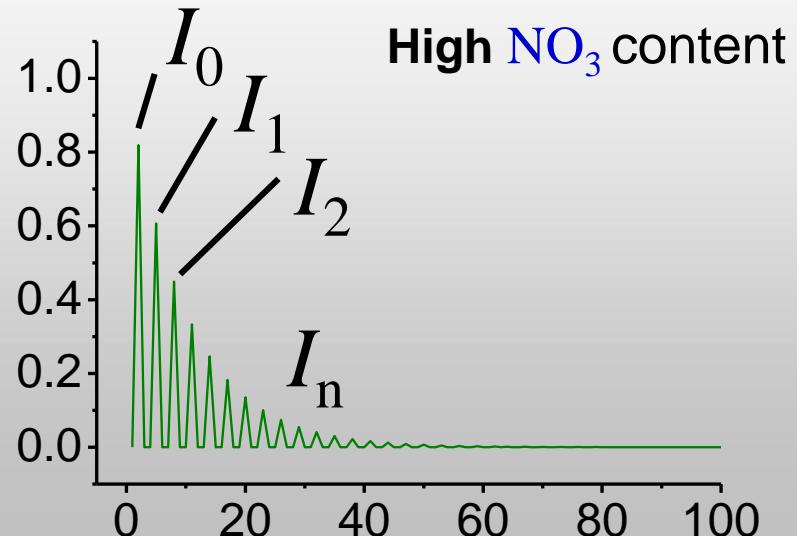
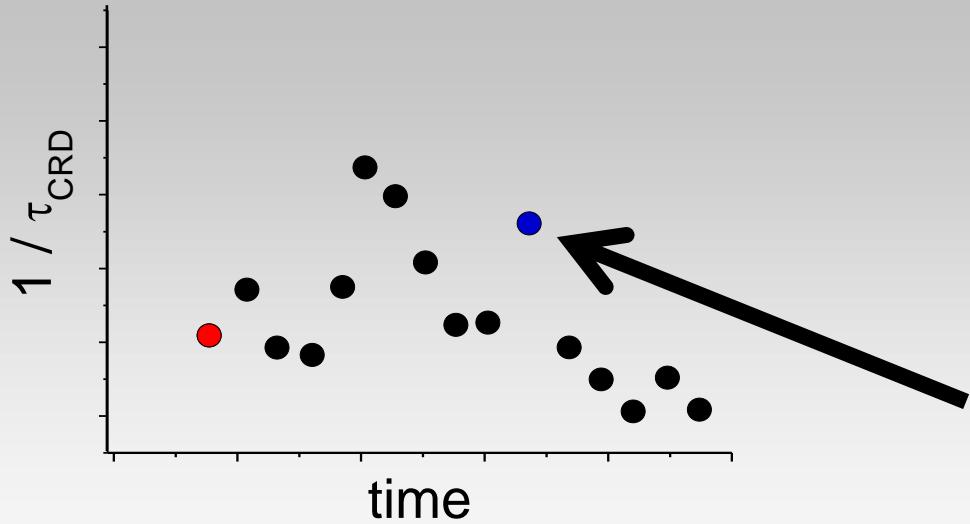
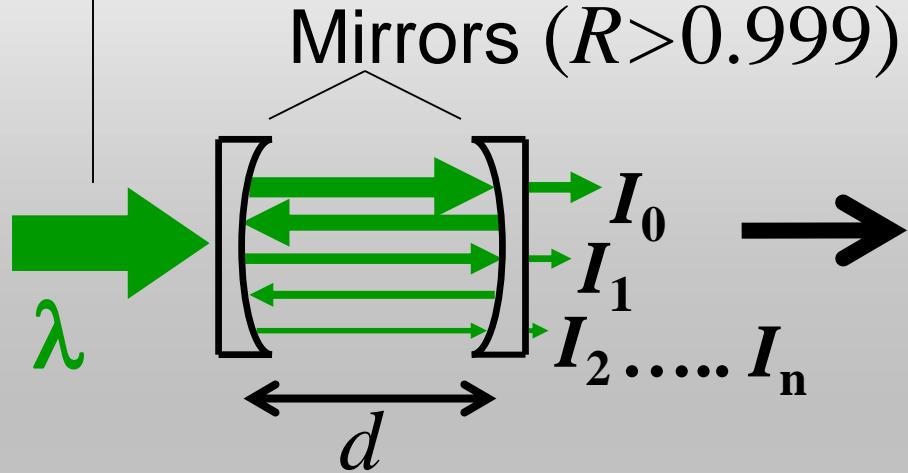
fit \downarrow

$$I(t) = I_0 \exp\left(-\frac{t}{\tau_{\text{crd}}}\right)$$

$$\tau_{\text{crd}}^{-1} = \frac{(1-R) c}{d} + c \sigma \underline{n_{\text{NO}_3}}$$

Principle of the cavity ring-down method

Light pulse

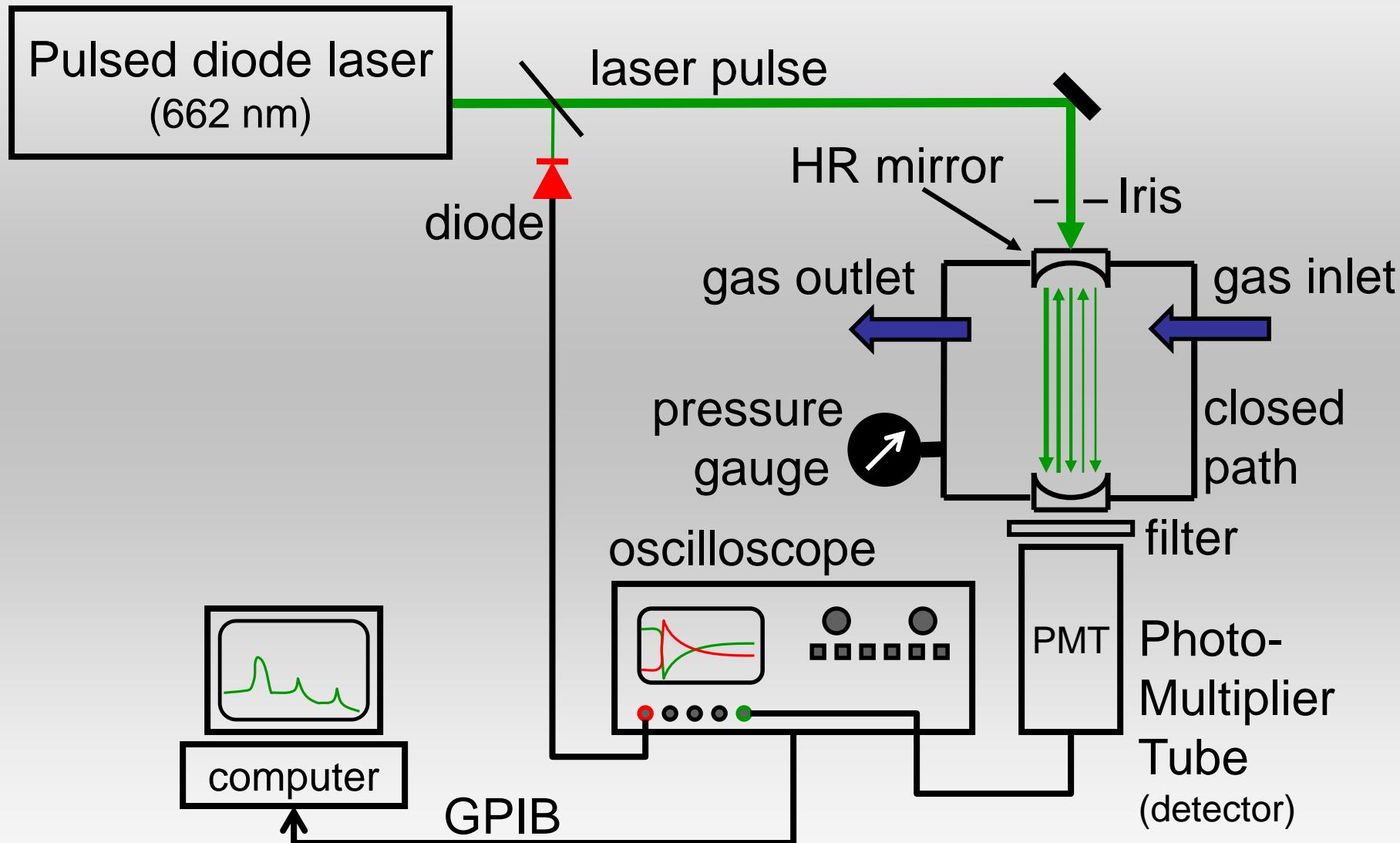


fit \downarrow

$$I(t) = I_0 \exp\left(-\frac{t}{\tau_{\text{crd}}}\right)$$

$$\tau_{\text{crd}}^{-1} = \frac{(1 - R) c}{d} + c \sigma \underline{n_{\text{NO}_3}}$$

Cavity Ring-Down Experiment



Objectives / Activities (3)

- Setup a vacuum tight, pressure controlled, *closed-path* cavity.
- Setup software for measurement and data analysis.
- Measure ring-down times in the empty cavity to characterize the optical losses (e.g. as a function of pressure).
- Detect H_2O absorption in the cavity as a function of time (optimize integration time).
- Synthesize NO_3 ($=\text{NO}_2 + \text{O}_3$) in the cavity and detect the nitrate radical concentrations as a function of time (optimize integration time).